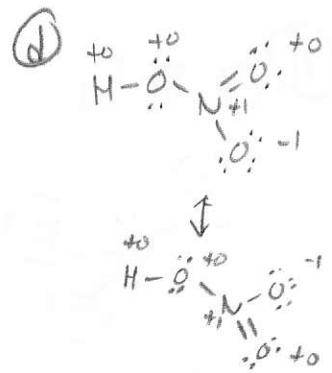
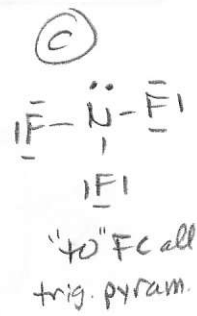
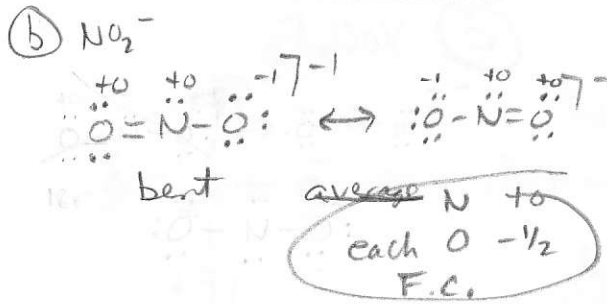
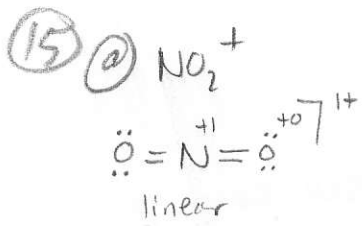
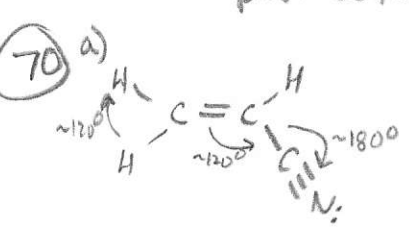
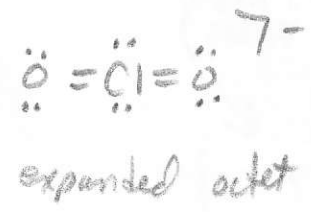
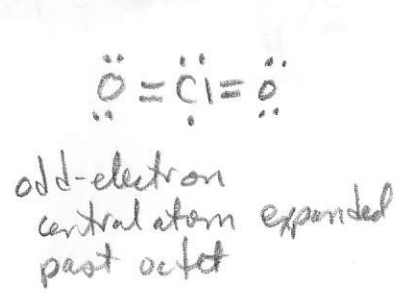
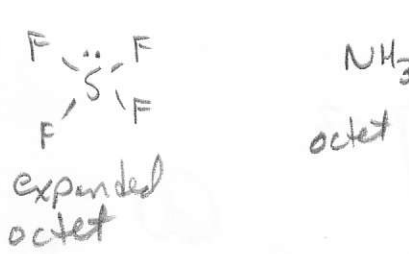
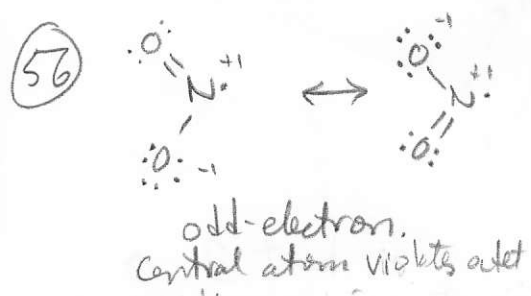
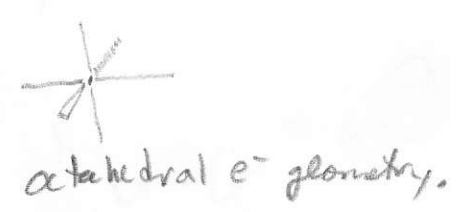
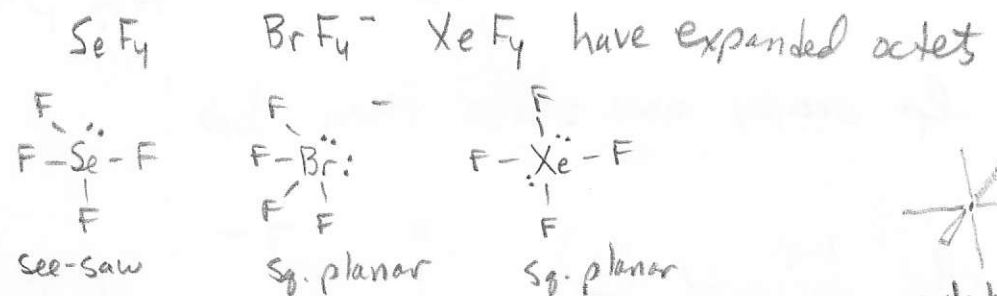


Ch 8



55) octet? BF_4^- SiF_4 have octets



- b, c) $\text{C}=\text{C}$ shorter & stronger than $\text{C}-\text{C}$
- d) positive regions are near the H-atoms. Neg Region near the N
- e) $\text{C}\equiv\text{N}$ bond is most polar
- f) yes, the molecule is polar

